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Metabolic and photosynthetic consequences of blocking starch biosynthesis in the green alga *Chlamydomonas reinhardtii* sta6 mutant

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SUMMARY

Upon nutrient deprivation, microalgae partition photosynthate into starch and lipids at the expense of protein synthesis and growth. We investigated the role of starch biosynthesis with respect to photosynthetic growth and carbon partitioning in the *Chlamydomonas reinhardtii* starchless mutant, sta6, which lacks ADP-glucose pyrophosphorylase. This mutant is unable to convert glucose-1-phosphate to ADP-glucose, the precursor of starch biosynthesis. During nutrient-replete culturing, sta6 does not re-direct metabolism to make more proteins or lipids, and accumulates 20% less biomass. The underlying molecular basis for the decreased biomass phenotype was identified using LC–MS metabolomics studies and flux methods. Above a threshold light intensity, photosynthetic electron transport rates (water → CO2) decrease in sta6 due to attenuated rates of NADPH re-oxidation, without affecting photosystems I or II (no change in isolated photosynthetic electron transport). We observed large accumulations of carbon metabolites that are precursors for the biosynthesis of lipids, amino acids and sugars/starch, indicating system-wide consequences of slower NADPH re-oxidation. Attenuated carbon fixation resulted in imbalances in both redox and adenylate energy. The pool sizes of both pyridine and adenylate nucleotides in sta6 increased substantially to compensate for the slower rate of turnover. Mitochondrial respiration partially relieved the reductant stress; however, prolonged high-light exposure caused accelerated photoinhibition. Thus, starch biosynthesis in *Chlamydomonas* plays a critical role as a principal carbon sink influencing cellular energy balance however, disrupting starch biosynthesis does not redirect resources to other bioproducts (lipids or proteins) during nutrient-replete culturing, resulting in cells that are susceptible to photochemical damage caused by redox stress.

Keywords: *Chlamydomonas reinhardtii*, sta6, photosynthetic electron transport chain, oxygen evolution rate, CBB cycle, malonyl CoA, NADPH re-oxidation.

INTRODUCTION

Microalgal strains partition photosynthetically fixed carbon into terminal products for biosynthesis and storage, including proteins, lipids and starch, in amounts that vary appreciably by genus, species and culturing conditions (Benamotz et al., 1985; Becker, 1994; Subramanian et al., 2013). Together with growth rate and tolerance to environmental stresses, these traits determine their suitability for biotechnological applications. The search for natural strains that accumulate desired products has been augmented by genetic approaches that aim to create transgenetic lines that are better suited with respect to one or more of these desired traits. For example, in *Chlamydomonas reinhardtii* (Chlorophyceae; referred to as *Chlamydomonas* throughout), genetically blocking alternative competing pathways, including starch biosynthesis, increased lipid content by two- to eightfold relative to control strains in N-deficient media (Wang et al., 2009; Li et al., 2010a,b; Radakovits et al., 2010; Work et al., 2010). The most widely characterized *Chlamydomonas* starchless mutant, sta6, contains a deletion in the small subunit of
ADP-glucose pyrophosphorylase (AGPase), which results in higher lipid production, and this mutant has been extensively investigated (Wang et al., 2009; Li et al., 2010a, b; Work et al., 2010; Goodson et al., 2011; Siaut et al., 2011; Fan et al., 2012; Blaby et al., 2013). It was found that, following growth of sta6, removal of all nitrogen and addition of acetate to the medium led to cells so engorged with lipids that they floated to the surface (Goodson et al., 2011; Goodenough et al., 2014). However, sta6 cells have been reported to display decreased photosynthetic activity compared to wild-type cells under some autotrophic and mixotrophic conditions (Li et al., 2010a; Work et al., 2010). Notably, sta6 displayed an altered energy partitioning at photosystem II (PSII), and a reduced photochemical yield that was attributed to lower Fv/Fm [a chlorophyll fluorescence parameter representing the maximum quantum efficiency of PSII photochemistry under continuous illumination] and qP (approximating the proportion of PSII reaction centers that are open under a given light intensity) relative to the reference strains (Li et al., 2010b), indicating that the starless phenotype exhibits consequences as far upstream as water oxidation by PSII. The reduced levels of water oxidation and CO2 fixation in sta6 renders this strain undesirable from an overall bioenergy perspective. To date, identification of metabolic bottlenecks induced by the absence of starch accumulation, and precise mapping of photosynthetic electron transport (PET) chain alterations in sta6, have not been performed; such aspects are explored in the present study.

Deletion of AGPase in sta6 eliminates the conversion of glucose–1–phosphate to ADP-glucose, the activated substrate used in covalent glucose linkage reactions. In the majority of prokaryotes, AGPase is a homotetrameric protein, but it is a heteromeric enzyme in eukaryotes (Ballicora et al., 2004), with separate small and large subunits. However, in both cases, AGPase targets glucose for incorporation into carbon and energy storage polysaccharides. In Arabidopsis, deletion of the small subunit of AGPase led to a 33% reduction in photosynthetic O2 evolution, while loss of the large subunit resulted in a moderate decrease of 8% (Sun et al., 1999). Similarly, in cyanobacterial mutants lacking AGPase, photosynthetic O2 evolution is highly inhibited (Suzuki et al., 2010; Grundel et al., 2012), indicating the importance of AGPase activity not only for producing starch/glycogen but also for maintaining photosynthetic activities by providing photosynthetic ATP and NADPH sinks.

Here, we have used a highly quantitative metabolomics approach to determine changes in the intracellular pool sizes of key metabolites, adenylate energy charge, and pyridine nucleotide redox poise. We combine this approach with measurements of proteins, lipids, carbohydrates, biomass and PET to identify the metabolic and photosynthetic bottlenecks in the production of energy carriers and carbon precursors for starch and lipid biosynthesis in sta6 and the starch-accumulating control strain cw15.

RESULTS

Photoautotrophic growth and biomass accumulation

Figure 1(a) compares sta6 and cw15 growth rates in photoautotrophic HS medium (see Experimental procedures) supplemented with 5 mM NaHCO3, illuminated using white light (100 μE m–2 sec–1). The sta6 mutant grew more slowly relative to cw15, with a doubling time that was approximately 30% longer (P = 0.03). Both strains reached stationary phase simultaneously, but sta6 had an approximately one-third lower cell density relative to the control strain. As the cultures grew, the light penetration decreased because of self-shading, with transmitted intensity dropping to as low as 25 μE m–2 sec–1 in stationary phase cultures.

At stationary phase (after 96 h of growth), the three major biochemical fractions (total reducing carbohydrate, lipids and proteins) were quantified in sta6 and cw15 (Figure 1b and Table 1). In addition, we also measured the biomass distribution in two other starch-synthesizing reference strains (CC124 and D66) to assess biomass distribution across multiple control strains. The dry cell weights of the sta6 mutant cultures were approximately 20–35%
lower than the dry cell weights of all three reference strains. The total reducing carbohydrate (TRC) content in sta6 was between two and five times lower than that in the control strains. Additionally, sta6 has a slightly increased protein content relative to D66 and CC124; however, this is not statistically different when compared to cw15.

These results indicate that loss of starch biosynthesis does not result in the repartitioning of carbon from TRC to lipids and proteins under nutrient-replete conditions; instead cellular biomass is reduced by a degree that is similar to the reduction in TRC levels in sta6 relative to the control strains. The decreased biomass accumulation implies that a primary adaptation in this starchless mutant is decreased/altered photosynthetic electron transport.

**Photosynthetic oxygen evolution rate, PSII quantum yield of charge separation, and electron transport rate**

To quantify the effect of a lack of starch biosynthesis on photosynthesis, photosynthetic O₂ evolution rates (OERs) were measured for sta6 and cw15 using a Clark electrode after illumination with a range of light intensities ($i_{\text{max}} = 620$ nm). At light levels up to 60 µE m⁻² sec⁻¹, sta6 and cw15 have comparable OER, but as the light intensity increased above this threshold, the differences between cw15 and sta6 increased, and at a light intensity of 680 µE m⁻² sec⁻¹, sta6 shows a 3.5-fold lower OER than cw15 (Figure 2a). It important to note that cultures were grown using white light at 60–80 µEm⁻² sec⁻¹, and then OERs were assayed across a range of light intensities. The lower OER in sta6 (Figure 2a) indicates that either PSII itself is impaired and/or that reactions downstream of PSII are affected at light intensities >60 µE m⁻² sec⁻¹. To differentiate between the two processes, we assessed the light-induced electron transport rate (ETR) through PSII and the quantum yield of PSII charge separation and the turnover efficiency of the water-oxidizing complex (WOC). The light-induced ETR through PSII was measured using a pulse-amplitude modulated fluorometer (Schreiber et al., 1997) (Figure 2b). This method indicates the proportion of open PSII centers capable of photoreducing Qₐ as a function of the light intensity of concurrent background illumination. Although widely used to measure ETR (flux), this signal is

![Figure 2](image_url)

*Figure 2.* Oxygen evolution rate, ETR and NPQ. (a) Oxygen evolution rates at various actinic light intensities (16, 30, 61, 100, 150, 340 and 680 µE m⁻² sec⁻¹) measured (at mid-log phase growth, Figure 1). (b) Relative electron transport rate (ETR). (c) NPQ measured by a pulse-amplitude modulated fluorometer. Values are means and standard errors of three biological replicates.

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Table 1. Biomass composition of sta6 and three reference strains, cw15, CC124 and D66, under NH₄Cl-replete photoautotrophic culturing conditions

<table>
<thead>
<tr>
<th></th>
<th>sta6</th>
<th>cw15</th>
<th>CC124</th>
<th>D66</th>
</tr>
</thead>
<tbody>
<tr>
<td>DW (0 h)</td>
<td>0.13 ± 0.04</td>
<td>0.07 ± 0.00</td>
<td>0.18 ± 0.03</td>
<td>0.12 ± 0.04</td>
</tr>
<tr>
<td>DW (96 h)</td>
<td>1.08 ± 0.05</td>
<td>1.34 ± 0.13</td>
<td>1.29 ± 0.11</td>
<td>1.49 ± 0.06</td>
</tr>
<tr>
<td>DW flux (g l⁻¹ d⁻¹)</td>
<td>0.23 ± 0.02</td>
<td>0.32 ± 0.02</td>
<td>0.27 ± 0.02</td>
<td>0.34 ± 0.03</td>
</tr>
<tr>
<td>Lipid (96 h)</td>
<td>0.072 ± 0.016</td>
<td>0.059 ± 0.010</td>
<td>0.106 ± 0.008</td>
<td>0.109 ± 0.023</td>
</tr>
<tr>
<td>% Lipid</td>
<td>6.6 ± 1.2</td>
<td>4.4 ± 0.3</td>
<td>8.3 ± 1.2</td>
<td>7.3 ± 1.5</td>
</tr>
<tr>
<td>TRC(96 h)</td>
<td>0.106 ± 0.006</td>
<td>0.428 ± 0.072</td>
<td>0.246 ± 0.029</td>
<td>0.493 ± 0.142</td>
</tr>
<tr>
<td>% TRC</td>
<td>9.88 ± 0.14</td>
<td>31.8 ± 2.4</td>
<td>19.1 ± 1.2</td>
<td>33.0 ± 8.8</td>
</tr>
<tr>
<td>Protein (96h)</td>
<td>0.688 ± 0.153</td>
<td>0.761 ± 0.293</td>
<td>0.645 ± 0.109</td>
<td>0.702 ± 0.083</td>
</tr>
<tr>
<td>% Protein</td>
<td>62.4 ± 18.0</td>
<td>56.1 ± 16.5</td>
<td>49.9 ± 7.2</td>
<td>47.5 ± 7.3</td>
</tr>
<tr>
<td>Mass balance</td>
<td>80.5 ± 16.3</td>
<td>92.2 ± 19.2</td>
<td>77.2 ± 8.7</td>
<td>87.8 ± 6.0</td>
</tr>
</tbody>
</table>

Values are reported as g l⁻¹, and values are means ± SD for three independent biological replicates.
a static measure of the QA population. It is proportional to the sum of ETR flux and photochemical quenching that does not lead to electron transfer beyond PSII. For this reason, the pulse-amplitude modulated measurement of ETR includes non-flux contributions to photochemical quenching within PSII that produce no O₂. The ETR curves show a trend that is qualitatively similar to that for OER, with sta6 and cw15 showing no significant differences at the lowest light intensities, but with differences between sta6 and cw15 emerging at approximately 75 μE m⁻² sec⁻¹. sta6 shows substantially lower ETRs at light intensities above 200 μE m⁻² sec⁻¹. At the maximum light intensity used for the measurement (approximately 400 μE m⁻² sec⁻¹), the ETR in sta6 was approximately twofold lower than that in cw15. The ETR in sta6 is also saturated at a lower light intensity (200 μE m⁻² sec⁻¹). The percentage differences observed in OER are larger than the differences in ETR (Figure S1) at all light intensities, and the difference becomes larger as the intensity increases. This comparison shows that light-induced O₂ consumption reactions contribute to ETR more significantly in sta6 than cw15. To investigate the possibility of non-photochemical quenching (NPQ)-dependent decreases in ETR, NPQ was quantified simultaneously with ETR using the method described by Schreiber et al. (1997), and calculated as described in Experimental procedures. sta6 and cw15 do not show significant differences in NPQ (Figure 2c), and thus the differences in ETR are not due to differences in excitation quenching in the antenna.

To examine whether an impaired PSII WOC activity is the cause of the low OER and ETR phenotype, the quantum yield of PSII charge separation and flux through the WOC were measured as chlorophyll variable fluorescence emission (Fv/Fm) using a fast repetition rate fluorometer at two flash frequencies (Ananyev and Dismukes, 2005). Compared to cw15, sta6 exhibits a slightly lower steady-state dark-adapted Fv/Fm under both a low flash frequency (4 Hz, <Fv/Fm> approximately 0.43-0.41) and a high flash frequency (100 Hz, <Fv/Fm> approximately 0.24-0.18) (Figure 3a). Following an initial dark adaptation, the rate of damping of the period four oscillations in Fv/Fm was similar for both strains, and decreased at higher flash rates (Figure 3a), indicating that both strains have normal initial WOC activity.

Fitting of the oscillations to an advanced Kok model (VZAD algorithm) (Vinyard et al., 2013) produce almost identical hit parameters (γavg) and periods (P) at both flash frequencies, indicating no significant difference in the initial turnover probability of the WOC in the two strains. However, the initial rate of damping of individual oscillations in Fv/Fm decreases rapidly with a higher flash rate (γavg decreases to 0.57 at 100 Hz, Figure 3a), and the non-oscillating portion of Fv/Fm in sta6 lacks the portion attributed to filling of the PQ pool (hatched area in Figure 3a) (Kolling et al., 2009). Thus, sta6 is characterized by

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**Figure 3.** Variable fluorescence, metabolic inhibitors and electron transport chain acceptors. (a) Mean chlorophyll variable fluorescence yield (Fv/Fm) from PSII produced by 50 single-turnover laser flashes at two flash rates (4 and 100 Hz). The mean Kok hit parameter, γavg, was determined by fitting the data to the VZAD model: γavg = 1 – α (miss) – β (double hit) – δ (backward) – ε (inactivated). The cycle period, P, was determined by Fourier transformation. The shaded region represents the reduced PQ pool. (b) OER of cw15 and sta6 in the presence of inhibitors to the alternative electron flow pathway at 680 μE m⁻² sec⁻¹. The control contains 5 mM bicarbonate, salicylhydroxamic acid (SHAM, 20 μM) inhibits alternative oxidase, propyl gallate (PGAL, 2 mM) inhibits plastoxinol terminal oxidase, and myxothiazol (2 μM) inhibits respiration. (c) Light-saturated OERs (680 μE m⁻² sec⁻¹) in the presence of artificial electron acceptors (40 μM 2,5-dichloro-4-benzoquinone, 400 μM N-dimethyl-4-nitrosoaniline or 5 mM NaHCO₃). Values are means and standard errors of three biological replicates.
a slower rate of emptying of the reduced plastoquinol PQH₂ pool relative to cw15.

Thus, both strains have a normal functioning WOC and PSII charge separation, but sta6 has an inhibited PQH₂ oxidation rate, reflecting a blockage in electron flow downstream of the PQH₂ pool. This blockage is responsible for the sharp break in OER flux at a light intensity >100 µE m⁻² sec⁻¹ (Figure 2a).

**Alternative oxygen-consuming pathways**

As previously mentioned, at higher light intensities, there is an imbalance in the extent of inhibition indicated by OER versus ETR, with OER predicting a lower intensity threshold and larger activity gap at saturation between sta6 and cw15 (Figure S1). This difference indicates the presence of alternative O₂-consuming pathways that influence OER and ETR differently at light intensities above 100 µE m⁻² sec⁻¹. The main pathways that consume O₂ are respiration and alternative oxidases in the mitochondrion, as well as plastoquinol terminal oxidase, the Mehler reaction and photorespiration in the chloroplast. Inhibition of alternative oxidases and plastoquinol terminal oxidase (also known as chlororespiration) by the specific inhibitors salicylhydroxamic acid (20 µM) and propyl gallate (2 mM), respectively, did not have any significant effect on the OER (at 680 µE m⁻² sec⁻¹) in either cw15 or sta6 (Figure 3b). However, inhibition of mitochondrial respiration by myxothiazol (inhibiting the cytochrome bc₁ complex of the respiratory chain) leads to a 70% reduction of the OER in sta6, but only a minor reduction (10%) in cw15 (Figure 3b) (at 680 µE m⁻² sec⁻¹). Thus, only in sta6 does mitochondrial respiration act as an important shunt to partially consume the excess light-induced reductant generated in the chloroplast. Further measurements of the Mehler reaction and photorespiration were not performed, although Figure S1 clearly indicates their possible contributions to re-oxidation.

**Normal PET rates through PSII and PSI may be restored in sta6**

The above results suggest that, under high light intensities, photosynthesis in sta6 is limited by the reactions occurring downstream of PSII, either within the electron transport chain itself (light-dependent reactions) or by carbon fixation reactions (light-independent reactions). To distinguish between these options, artificial electron acceptors were used to alleviate bottlenecks in the PET chain, and the corresponding OER was measured at 680 µE m⁻² sec⁻¹. As shown in Figure 2(c), in the presence of 2,5-dichloro-p-benzoquinone (40 µM), which accepts electrons from the Q₀ site of PSII, the OER of both sta6 and cw15 increased to approximately 200 µmol O₂ mg chlorophyll⁻¹ h⁻¹. In the presence of N-dimethyl-4-nitrosoaniline (400 µM), which accepts electrons from photosystem I (PSI) via ferredoxin, both strains had an equal OER of approximately 110 µmol O₂ mg chlorophyll⁻¹ h⁻¹. These experiments show that sta6 has a structurally intact electron transport chain similar to that of cw15 through ferredoxin, with normal fluxes and similar electron acceptor pool sizes (for PQ and ferredoxin). Bicarbonate addition to ensure saturating levels for carbon fixation reactions (Figure 3b) did not reverse the disparity in OER between cw15 and sta6. Thus, the OER in sta6 is limited downstream of ferredoxin by carbon fixation reactions.

To further isolate the location of flux bottlenecks, we measured the light-induced yield of O₂ from a thin solution layer of cells (0.8 mm) illuminated at 400 µE m⁻² sec⁻¹ using a bare platinum (rate) electrode. This method resolves kinetic features of the electron acceptor pools downstream of PSII (Zakrzhesvskii et al., 1978). A typical O₂ evolution trace after dark adaptation displays three distinct time-resolved features. The first is a sharp transient peak from reduction of the PQ pool, the second is a broad peak arising from reduction of NADP⁺, and the third is a continuous slope arising from CO₂-dependent O₂ evolution. The initial peak has a rise time of 300 msec due to O₂ diffusion to the electrode, and is determined by the finite thickness of the sample. The area under this peak (see Experimental procedures) is directly proportional to the difference in electron fluxes filling and emptying the oxidized PQ pool. The PQ peak intensity and area are indistinguishable for sta6 and cw15. In cw15, a second O₂ feature appears as an inflection (Figure 4, top), but this is masked by a much larger continuous slope that disappears when CO₂ is removed (data not shown) or when the PET inhibitor 3-(3,4-dichlorophenyl)-1,1-dimethyleurea is added (Figure 4, bottom). This CO₂-dependent water oxidation rate is sevenfold greater in cw15 than in sta6 over the illumination period shown (400 µE m⁻² sec⁻¹).

To quantify the transient flux into and out of the NADP (H) pool, the relative light-induced yield of the reaction PSI donor → NADP⁺ was monitored by the fluorescence change from reduced NAD(P)H in vivo. NAD(P)H kinetics in whole cells were measured using a home-built fluorometer (excitation at 365 nm, detection at 420 nm) similar to that described by Mi et al. (2000). Samples were exposed to a dark/light/dark regime of 5 min/4 min/7 min using saturating light (400 µE m⁻² sec⁻¹, 660 nm). Curves derived from three biological replicates for sta6 and two for cw15 are shown in Figure 5(a), with results for individual replicates shown in Figure S2A. The rise and decay times and the amplitude changes of the fluorescence yield are given in Table 2. Light-induced changes in NAD(P)H fluorescence display dynamics similar to those described in previous reports (Cerovic et al., 1993; Mi et al., 2000). We observed an increase in NADPH fluorescence emission during actinic illumination, signifying light-driven reduction of NADP⁺, followed by a steady-state phase during which the NADP⁺
reduction and NAD(P)H oxidation rates are matched. After illumination (i.e. in darkness), the fluorescence decays, followed by recovery to the initial steady-state dark level. In the presence of the artificial PSI electron acceptor N-dimethyl-4-nitrosoaniline, both sta6 and cw15 show no change in NAD(P)H fluorescence upon illumination (Figure S2B).

The NAD(P)H fluorescence yield increases from the dark-adapted level to the maximal photoreduced level over identical time periods for both strains (t1 = 7.2 sec). During the illumination period, the larger photoinduced amplitude in sta6 decreases slowly to 50% with a lifetime of approximately 2.9 min (t2), but the lower photoinduced amplitude for cw15 does not decay appreciably over the measurement time scale. The larger amplitude and gradual decaying slope of light-induced NADPH in sta6 are attributed to slower NADPH re-oxidation. This was verified by turning off the actinic light and monitoring the decay rate of the NADPH fluorescence amplitude (Y2), which is indeed slower in sta6 (t3 = 20 sec) than cw15 (t3 = 14 sec), and ‘overshoots’ to a level lower than the initial steady-state dark level. The t3 decay rate in cw15 is followed by recovery to the initial steady-state level within 39 sec (t4), as the non-equilibrium level of oxidized NADP+ is re-reduced in the dark. Such oscillating behavior is commonly associated with regenerative cycles. This latter phase is absent in sta6. Taken together, the lack of overshoot, slower t3 decay, and slow decay of the actinic phase (t2) all support the conclusion of a slower re-oxidation rate of actinically generated NADPH in sta6. For cw15, the lack of observable decay of the actinically generated NADPH phase indicates a rate of NADPH re-oxidation faster than the time resolution of the instrument (approximately 1 sec).

Figure 4. Light-saturated OERs (400 µE m⁻² sec⁻¹) on a bare platinum electrode recorded as nanoAmperes (µg chlorophyll)⁻¹ in the presence of bicarbonate as the sole electron acceptor. Values are means of three biological replicates. Light was turned on or off at the points indicated by solid arrows. Transient O₂ evolution shows two distinct phases: spike 1 represents the PQ pool reduction, while the continuous slope represents carbon fixation (top panel). The inset shows the PQ pool size equivalent calculated as described in Experimental procedures. The bottom panel shows the effect of 10 µM 3-(3,4-dichlorophenyl)-1,1-dimethylurea (DCMU) on the OER. The dotted line represents the current recorded in the absence of illumination.

Figure 5. NAD(P)H reoxidation and photoinhibition. (a) Effect of actinic illumination on induction of a continuous level of blue/green fluorescence arising from NAD(P)H. Actinic illumination was provided by saturating red light (660 nm, intensity = 400 µE m⁻² sec⁻¹). Light was turned on or off at the points indicated by solid arrows. The fluorescence decay during illumination (lifetime indicated by t3 for sta6) was fitted to a linear decay slope and extrapolated to ‘light on’ time to calculate the maximal photoreduced NADPH (Y1) in the absence of oxidation of NADPH. Y2 denotes the oxidized NADPH post-illumination. t1, t2, t3 and t4 are the rise and decay times as described in the text. A magnification of the decay kinetics is shown in the inset. The results for sta6 are offset by 3 units in (a). (b) Photoinhibition at 400 µE m⁻² sec⁻¹, the t3 values for photoinhibition are given in the inset table. Values are means and standard errors of three biological replicates.

Table 2  Kinetics of NADP⁺ reduction and NADPH re-oxidation

<table>
<thead>
<tr>
<th>Parameters</th>
<th>cw15</th>
<th>sta6</th>
<th>Fold change (sta6/cw15)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Rise time, t₁ (sec)</td>
<td>7.2 ± 0.3</td>
<td>7.2 ± 0.42</td>
<td>1.0 ± 0.67</td>
</tr>
<tr>
<td>Re-oxidation time, t₂ (min)</td>
<td>≤ t₁</td>
<td>2.9 ± 0.67</td>
<td></td>
</tr>
<tr>
<td>Decay time, t₃ (sec)</td>
<td>14 ± 0.6</td>
<td>20 ± 2.4</td>
<td>1.4 ± 0.12</td>
</tr>
<tr>
<td>Re-reduction rate, t₄ (sec)</td>
<td>39 ± 0.6</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Maximum photoreduced NADPH, Y₁ (relative units)</td>
<td>1.5 ± 0.04</td>
<td>2.6 ± 0.3</td>
<td>1.7 ± 0.12</td>
</tr>
<tr>
<td>Oxidized NADPH post-illumination, Y₂ (relative units)</td>
<td>2.8 ± 0.3</td>
<td>1.9 ± 0.2</td>
<td>0.68 ± 0.15</td>
</tr>
</tbody>
</table>

Values are means ± SEM for three biological replicates.

**sta6 is more sensitive to photoinhibition**

Sensitivity to photoinhibition upon prolonged high-light exposure was determined by measuring the loss of PSII quantum yield (as F₇₀/Fₐₐₚ) in samples incubated under light (400 μE m⁻² sec⁻¹, 620 nm) over a period of 6 h (Figure 5b). sta6 exhibits a 35% faster photoinhibition rate than in cw15 (half life = 0.28 h versus 0.43 h), indicating the importance of starch biosynthesis for tolerance to high-light stress.

**sta6 is inefficient in drain upper glycolytic intermediates at high light intensity**

Given that OER and ETR are significantly inhibited at high light intensities and the metabolic block occurs downstream of PET, comparative metabolic analysis of samples treated under low light (LL) or high light (HL) was performed to reveal the metabolic consequences of the decrease in NADPH re-oxidation. The pool sizes of various metabolites involved in central carbon metabolism were measured using LC-MS/MS (Bennette et al., 2011) under ambient growing conditions (approximately 60–80 μE m⁻² sec⁻¹, low light) and after a 2 h exposure to high light (400 μE m⁻² sec⁻¹). Concentrations of the metabolites are shown in Table 3. The results for metabolites of the Calvin–Benson–Bassham (CBB) cycle and upper glycolysis are summarized in Figure 6(a,b). Simultaneously, the biomass contents (lipids, TRC and protein) were also measured (Figure 6c) using the same cultures.

Upon transfer to HL, neither protein nor lipids showed a significant change in the two strains. However, the TRC content per cell increased by 25% in cw15 but was unchanged in sta6 (Figure 6c). Thus, under 2 h exposure to HL, fixed carbon predominantly accumulates as starch, whereas blocking starch biosynthesis at AGPase in sta6 impedes carbon assimilation into all biopolymers.

Metabolite analysis revealed that, under HL, cw15 shows an increase in CBB cycle intermediates including ribulose-1,5-bisphosphate (fivefold), ribose-5-phosphate (fourfold) and xylulose-5-phosphate (ninefold) compared to LL. Glycer-aldehyde phosphate and 3-phosphoglycerate levels also increase by approximately twofold (Figure 5a and Table 3). This increase correlates with the enhanced carbon assimilation in cw15 (Figure 6c). Even though the TRC content increases in cw15 (Figure 6c), there is no increase in the pool size of upper glycolytic intermediates, indicating an efficient drain of the upper glycolytic metabolites into carbohydrate biosynthesis. By contrast, in sta6, we observed large increases in the levels of many CBB cycle metabolites, especially the glyceraldehyde phosphate and 3-phosphoglycerate levels, which increased by eleven- and sevenfold, respectively. Additionally, we observed increased levels of the upper glycolytic intermediates glucose-1-phosphate, glucose-6-phosphate and fructose-6-phosphate (fivefold, fourfold and eightfold, respectively), even though there is no significant increase in the TRC content. Thus, under HL conditions and in the absence of AGPase, there is an inefficient drain of the upper glycolytic intermediates that normally flow into starch biosynthesis.

A lower ATP recycling rate may limit electron transport chain flux (Sharkey et al., 1986). To test this, the adenine nucleotide concentrations were quantified using LC-MS, and the cellular energy charge was calculated (Table 3). sta6 shows appreciably higher concentrations of all adenine nucleotides (ATP, ADP and AMP) in sta6 relative to cw15 under both LL and HL. Additionally, sta6 has a marginally larger pool size of the phosphorylated pyridine nucleotides relative to cw15 (Table 3). This increase in both NADP(H) and adenylates probably reflects the cellular response to sta6’s inability to use the CBB intermediates in starch biosynthesis. Although the cellular energy charge did not differ in the two strains under either light regime, the method we used may not be rapid enough to prevent equilibration. A stable cellular energy charge is known to be essential for normal cellular function (Kramer and Evans, 2011).

**Photosynthetically fixed carbon in sta6 is re-directed towards the lower glycolytic pathway**

Metabolites of the central carbon metabolism outside the CBB cycle and involved in biosynthesis of amino acids, fatty acids and alternative CO₂ fixation (C4) were quantified and compared across the two strains (Table 3). In sta6, levels of the high-energy phosphate ester phosphoenolpyruvate increased fourfold, further indicating accumulation of excess cellular phosphate and energy. Among the tricarboxylic acid cycle metabolites, succinate and 2-ketoglutarate increased uniformly in both strains. Although sta6 does not accumulate protein, the amino acid precursors glutamine and glutamate increased under both LL and HL compared to cw15. Significantly, from the perspective of lipid biosynthesis, the level of the fatty acid precursor...
Table 3 LC-MS/MS results for various metabolites

<table>
<thead>
<tr>
<th>Metabolite</th>
<th>Abbreviation</th>
<th>cw15 under LL</th>
<th>cw15 under HL</th>
<th>sta6 under LL</th>
<th>sta6 under HL</th>
<th>Fold change for cw15 (HL/LL)</th>
<th>Fold change for sta6 (HL/LL)</th>
<th>P value for cw15</th>
<th>P value for sta6</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Glycolysis and CBB intermediates</strong></td>
<td>ADP-glucose</td>
<td>0.019 (0.003)</td>
<td>0.010 (0.001)</td>
<td>0.001 (0.001)</td>
<td>0.001 (0.000)</td>
<td>0.515 (0.102)</td>
<td>0.675 (0.311)</td>
<td>0.043</td>
<td>0.470</td>
</tr>
<tr>
<td></td>
<td>G1P</td>
<td>0.115 (0.006)</td>
<td>0.086 (0.017)</td>
<td>0.063 (0.004)</td>
<td>0.357 (0.091)</td>
<td>0.745 (0.152)</td>
<td>5.63 (1.19)</td>
<td>0.179</td>
<td>0.022</td>
</tr>
<tr>
<td></td>
<td>G6P</td>
<td>0.101 (0.001)</td>
<td>0.076 (0.025)</td>
<td>0.064 (0.004)</td>
<td>0.291 (0.104)</td>
<td>0.753 (0.246)</td>
<td>4.53 (1.14)</td>
<td>0.373</td>
<td>0.046</td>
</tr>
<tr>
<td></td>
<td>F6P</td>
<td>0.058 (0.006)</td>
<td>0.055 (0.029)</td>
<td>0.038 (0.001)</td>
<td>0.313 (0.118)</td>
<td>0.950 (0.509)</td>
<td>8.27 (1.63)</td>
<td>0.927</td>
<td>0.050</td>
</tr>
<tr>
<td></td>
<td>GAP</td>
<td>0.077 (0.002)</td>
<td>0.195 (0.048)</td>
<td>0.040 (0.002)</td>
<td>0.474 (0.183)</td>
<td>2.51 (0.626)</td>
<td>11.8 (2.58)</td>
<td>0.058</td>
<td>0.041</td>
</tr>
<tr>
<td></td>
<td>3PG</td>
<td>3.00 (0.091)</td>
<td>5.69 (0.812)</td>
<td>0.366 (0.058)</td>
<td>2.87 (1.091)</td>
<td>1.89 (0.277)</td>
<td>7.46 (1.04)</td>
<td>0.030</td>
<td>0.035</td>
</tr>
<tr>
<td></td>
<td>PEP</td>
<td>0.736 (0.041)</td>
<td>1.27 (0.210)</td>
<td>0.160 (0.014)</td>
<td>0.702 (0.264)</td>
<td>1.72 (0.302)</td>
<td>4.38 (1.695)</td>
<td>0.067</td>
<td>0.110</td>
</tr>
<tr>
<td><strong>Acetyl CoA</strong></td>
<td>AcCoA</td>
<td>0.116 (0.003)</td>
<td>0.231 (0.030)</td>
<td>0.205 (0.014)</td>
<td>0.119 (0.042)</td>
<td>1.99 (0.263)</td>
<td>0.581 (0.209)</td>
<td>0.018</td>
<td>0.125</td>
</tr>
<tr>
<td><strong>Ribulose-1,5-bisphosphate</strong></td>
<td>RuBP</td>
<td>0.082 (0.011)</td>
<td>0.337 (0.019)</td>
<td>0.019 (0.002)</td>
<td>0.099 (0.037)</td>
<td>4.10 (0.538)</td>
<td>5.10 (0.906)</td>
<td>0.000</td>
<td>0.038</td>
</tr>
<tr>
<td><strong>Ribose-5-phosphate</strong></td>
<td>R5P</td>
<td>0.012 (0.002)</td>
<td>0.039 (0.006)</td>
<td>0.013 (0.002)</td>
<td>0.046 (0.016)</td>
<td>3.18 (0.680)</td>
<td>3.62 (1.37)</td>
<td>0.015</td>
<td>0.108</td>
</tr>
<tr>
<td><strong>Xylose-5-phosphate</strong></td>
<td>X5P</td>
<td>0.015 (0.003)</td>
<td>0.071 (0.026)</td>
<td>0.010 (0.002)</td>
<td>0.091 (0.031)</td>
<td>4.77 (2.08)</td>
<td>9.34 (3.76)</td>
<td>0.105</td>
<td>0.059</td>
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<tr>
<td><strong>Sedoheptulose-7-phosphate</strong></td>
<td>S7P</td>
<td>0.074 (0.022)</td>
<td>0.575 (0.145)</td>
<td>0.061 (0.017)</td>
<td>0.507 (0.174)</td>
<td>8.21 (3.6)</td>
<td>7.74 (3.01)</td>
<td>0.063</td>
<td>0.027</td>
</tr>
<tr>
<td><strong>Lipid precursor</strong></td>
<td>Malonyl CoA</td>
<td>0.001 (0.000)</td>
<td>0.001 (0.000)</td>
<td>0.018 (0.000)</td>
<td>0.028 (0.002)</td>
<td>0.968 (0.296)</td>
<td>1.59 (0.071)</td>
<td>0.9195</td>
<td>0.026</td>
</tr>
<tr>
<td><strong>Tricarboxylic acid cycle and amino acids</strong></td>
<td>Succinate</td>
<td>0.712 (0.058)</td>
<td>1.49 (0.131)</td>
<td>0.304 (0.044)</td>
<td>0.594 (0.164)</td>
<td>2.09 (0.251)</td>
<td>1.95 (0.608)</td>
<td>0.006</td>
<td>0.161</td>
</tr>
<tr>
<td><strong>Malate</strong></td>
<td>MAL</td>
<td>1.86 (0.108)</td>
<td>2.63 (0.201)</td>
<td>1.17 (0.152)</td>
<td>1.67 (0.457)</td>
<td>1.41 (0.136)</td>
<td>1.42 (0.429)</td>
<td>0.028</td>
<td>0.360</td>
</tr>
<tr>
<td><strong>α-ketoglutarate</strong></td>
<td>AKG</td>
<td>0.452 (0.017)</td>
<td>1.13 (0.334)</td>
<td>0.169 (0.030)</td>
<td>0.343 (0.096)</td>
<td>2.50 (0.746)</td>
<td>2.03 (0.675)</td>
<td>0.112</td>
<td>0.159</td>
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<tr>
<td><strong>Glutamate</strong></td>
<td>GLU</td>
<td>3.32 (0.177)</td>
<td>5.80 (0.378)</td>
<td>2.28 (0.106)</td>
<td>5.35 (1.29)</td>
<td>1.44 (0.147)</td>
<td>2.34 (0.578)</td>
<td>0.004</td>
<td>0.077</td>
</tr>
<tr>
<td><strong>Glutamine</strong></td>
<td>GLN</td>
<td>0.046 (0.002)</td>
<td>0.062 (0.001)</td>
<td>0.032 (0.019)</td>
<td>0.057 (0.018)</td>
<td>1.34 (0.071)</td>
<td>2.08 (0.276)</td>
<td>0.003</td>
<td>0.056</td>
</tr>
<tr>
<td><strong>Sugars</strong></td>
<td>Sucrose</td>
<td>0.011 (0.004)</td>
<td>0.014 (0.002)</td>
<td>0.007 (0.001)</td>
<td>0.044 (0.017)</td>
<td>1.20 (0.421)</td>
<td>6.10 (2.488)</td>
<td>0.598</td>
<td>0.062</td>
</tr>
<tr>
<td><strong>Energy carriers</strong></td>
<td>AMP</td>
<td>1.66 (0.171)</td>
<td>1.60 (0.830)</td>
<td>2.09 (0.521)</td>
<td>2.67 (1.19)</td>
<td>2.16 (0.656)</td>
<td>2.71 (1.329)</td>
<td>0.137</td>
<td>0.217</td>
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<tr>
<td></td>
<td>ADP</td>
<td>2.20 (0.206)</td>
<td>2.84 (0.363)</td>
<td>3.43 (0.398)</td>
<td>3.96 (2.23)</td>
<td>1.28 (0.231)</td>
<td>1.15 (0.572)</td>
<td>0.549</td>
<td>0.717</td>
</tr>
<tr>
<td></td>
<td>ATP</td>
<td>3.40 (0.183)</td>
<td>3.92 (0.326)</td>
<td>6.48 (1.35)</td>
<td>6.21 (1.21)</td>
<td>1.15 (0.256)</td>
<td>0.956 (0.285)</td>
<td>0.923</td>
<td>0.727</td>
</tr>
<tr>
<td><strong>Cellular energy charge</strong></td>
<td>ATP ADP AMP</td>
<td>0.682 (0.060)</td>
<td>0.637 (0.023)</td>
<td>0.620 (0.022)</td>
<td>0.638 (0.076)</td>
<td>1.92 (0.286)</td>
<td>3.06 (0.268)</td>
<td>0.043</td>
<td>0.752</td>
</tr>
<tr>
<td><strong>Pyridine nucleotides</strong></td>
<td>NAD+ + NADH</td>
<td>0.636 (0.052)</td>
<td>0.873 (0.073)</td>
<td>0.781 (0.027)</td>
<td>1.13 (0.262)</td>
<td>1.22 (0.088)</td>
<td>1.29 (0.246)</td>
<td>0.213</td>
<td>0.783</td>
</tr>
</tbody>
</table>

Values are means. Standard errors of the mean are given in parentheses. Bold numbers denote the metabolites with a 1.5-fold higher or 1.5-fold lower metabolic concentration under HL w.r.t LL. P-values are italicized.

Abbreviations: cw15 = cells with wild type genotype; sta6 = cells with the glucose-1,5-bisphosphate dehydrogenase mutants; HL = high light; LL = low light; LL = low light; cFold change in concentration of the metabolites between LL and HL in sta6 and cw15, calculated as μmol per 10^8 cells.
malonyl CoA is 18-fold higher in sta6 compared to cw15, and increased further to 28-fold under HL, although the total lipid content in sta6 was indistinguishable from that in cw15.

**DISCUSSION**

Various starchless mutants have been used as test strains in search of higher lipid producers in many previous studies (Wang et al., 2009; Li et al., 2010a,b; Work et al., 2010; Goodson et al., 2011; Siaut et al., 2011; Fan et al., 2012; Blaby et al., 2013; Goodenough et al., 2014). These studies also found that loss of starch synthesis and lower growth rates were strongly correlated, but the consequences on the individual PET reactions and carbon partitioning via pathways and into terminal products had not been examined systematically at the molecular level. Scheme 1 summarizes our data describing the molecular phenotypes arising from loss of AGPase in the Chlamydomonas starchless mutant sta6.

Starch biosynthesis (or glycogen biosynthesis in cyanobacteria) is essential for (at least) four reasons: (i) consuming the photosynthetically generated reductant...
(NADPH) and ATP (Kramer and Evans, 2011), (ii) providing a gluconeogenic route for replenishing CBB cycle intermediates, (iii) to produce a storage product with low osmotic potential to meet carbon and energy needs in the dark, and (iv) for photoprotection against oxidative damage. In the absence of starch biosynthesis arising from a non-functional AGPase, all of these functions are disrupted. Especially under high light, the primary effect is on the rate of CO2-dependent water oxidation, with sta6 having a 3.5-fold slower electron flux than cw15 (Figure 2a). This leads to a build-up in the reductant pool (NADPH/NADP+), the extent of carbohydrate phosphorylation, and the level of adenylate nucleotides, the latter presumably in an attempt to compensate for the slower turnover rate (Scheme 1). Accumulation of several phosphorylated carbohydrates (Table 3 and Scheme 1) and the upper glycolysis intermediates glucose-1-phosphate, glucose-6-phosphate and fructose-6-phosphate occurs, and these do not drain efficiently to terminal sinks. This outcome, together with the 2.8-fold larger adenylate pool size, results in a corresponding decrease in the free Mg2+ as well. As Mg2+ and free Pi are required activators for RuBisCO and other CBB enzymes, allostERIC down-regulation of RuBisCO dependent CO2 fixation activity is expected in sta6. This down-regulation appears to be directly responsible for the slowing of NADPH oxidation, and the ultimately slowing of all PET reactions as far upstream as water oxidation by PSII. The CBB cycle appears to be the main target affected by the AGPase mutation in sta6, as none of the individual PET enzyme fluxes exhibit inhibition relative to the cw15 control when measured in isolation. Our results with PET electron acceptors (Figure 3c) confirm that, like typical wild algal strains (Rochaix, 2011), re-oxidation of the PQ pool is retained in sta6 as the slowest kinetic step of the PET chain. sta6 uses mitochondrial respiration to help relieve aspects of cellular over-reduction (Figure 3b). By contrast, inhibition of mitochondrial respiration in cw15 leads to a comparatively insignificant effect on the OER. Thus, respiration appears to be critical for sustaining the low level of photosynthetic oxygen evolution in sta6. Consistent with this observation, P700+ kinetics are impaired when respiration is inhibited in sta6 (Johnson and Alric, 2012). For conditions under which sta6 is unable to maintain redox homeostasis, it does not depend on plastoquinol terminal oxidase or alternative oxidases but potentially uses the Mehler reaction or photorespiration to reduce O2 and remove the excess reductant. Thus alternative O2-consuming pathways become important for supporting light-induced ETR as reductves in sta6.

The large fourfold increase in phosphoenolpyruvate concentration in sta6 is interesting as this molecule is known to function in another CO2 fixation pathway via phosphoenolpyruvate carboxylase, which may therefore become more important in sta6. Phosphoenolpyruvate carboxylase catalyzes the reaction that combines bicarbonate (HCO3-) and phosphoenolpyruvate to form oxaloacetate. This is an anaplerotic reaction that is important for synthesis of amino acids (Ala, Asp, Glu). Phosphoenolpyruvate carboxylase-dependent carbon fixation may even exceed the amount of carbon fixed via the CBB cycle in some unicellular algae under nitrogen-limited conditions (Guy et al., 1989).
sta6 has smaller hexose phosphate pool sizes under low light, which may reflect the attenuation of photosynthesis, decreased carbon fixation, and/or re-direction of fixed carbon to the tricarboxylic acid cycle for protein synthesis. Under HL conditions, when increased levels of photosynthetic activity ensue, the tricarboxylic acid cycle appears to become saturated (Table 3), resulting in accumulation of hexose phosphate metabolites. Determination of the precise underpinnings of these phenotypes requires additional experimentation, and metabolic flux analysis is required to obtain a more informed understanding of the observed dynamic re-direction of metabolites. Even though over-reduction of the PET chain is relieved in part by mitochondrial respiration in sta6 (Figure 3b), prolonged exposure to high light leads to accelerated photo-inhibition relative to cw15 (Figure 5b), indicating that starch production is an important safeguard against photodamage. The light-driven accumulation of excess NADPH and reduced ferredoxin lead to formation of reactive oxygen species in many aquatic phototrophs and plants. A similar phenotype was observed in several other characterized AGPase mutants (Sun et al., 1999; Suzuki et al., 2010; Grundel et al., 2012). As starch is the least energy-intensive terminal product among starch, proteins and lipids, it may serve as an energy storage buffer to protect against light energy fluctuations and to enable biosynthesis of these more energy-intensive biopolymers during low light periods, similar to that postulated in higher plants (Caspar et al., 1985; Stitt and Quick, 1989; Ludewig et al., 1998; Geigenberger, 2011; Weise et al., 2011).

Carbon that enters the central metabolism from the CBB cycle may take two possible routes: the upper glycolytic pathway leading into C6 and C5 carbohydrates, or the lower glycolytic pathway leading to acetyl CoA that either enters the tricarboxylic acid cycle to synthesize amino acids or is converted to malonyl CoA to synthesize lipids. Several groups have reported increased lipid yields in sta6 after switching cells to a nitrogen-deficient growth medium (Wang et al., 2009; Li et al., 2010a,b; Work et al., 2010; Goodson et al., 2011; Siaut et al., 2011; Fan et al., 2012). Increased lipid accumulation is not observed under nutrient-replete, phototrophic conditions in sta6; instead growth is stunted. However, there is a major increase in both acetyl CoA and malonyl CoA levels in sta6. Malonyl CoA formation is catalyzed by acetyl CoA carboxylase (ACCase) through an ATP-dependent carboxylation of biotin that transfers a carboxyl group to acetyl CoA (Berg et al., 2002). The ACCase reaction is considered to be the committed step in fatty acid synthesis. It is regulated by reversible phosphorylation catalyzed by an AMP-dependent protein kinase (Berg et al., 2002). ACCase is inhibited by phosphorylation, while the unphosphorylated form has carboxylase activity. AMP-dependent protein kinase itself acts as an adenylate nucleotide sensor, being activated by high AMP levels and inhibited by high ATP levels (Berg et al., 2002; Hardie and Pan, 2002). As such, the kinase activity becomes self-limiting at high ATP concentrations. Accumulation of malonyl CoA in sta6 indicates a reversal of inhibition of ACCase relative to cw15. Given that ATP accumulates in sta6 (Table 3), we predict that the higher ATP content probably inhibits AMP-dependent protein kinase, and, in turn, enhances the carboxylase activity of ACCase. Thus, simple over-expression of ACCase may not be enough to enhance ACCase activity, but instead needs to occur in parallel with reducing the level of phosphorylation of ACCase, which is achieved by suppression of AMP-dependent protein kinase phosphorylation activity. Taken together, the higher ATP accumulation in sta6 cells metabolically poises it for higher levels of fatty acyl biosynthesis, as indicated by elevated levels of acetyl CoA and especially malonyl CoA, but downstream blockage in the fatty acid synthesis complex putatively prevents utilization. Our data show that factors beyond availability of the precursor (malonyl CoA) control the flux into lipid biosynthesis. Under phototrophic nutrient-replete conditions, downstream enzymatic reactions, possibly involving the fatty acid synthase complex or glycerolipid biosynthetic enzymes, appear to be limiting lipid biosynthesis. If the flux through these enzymes were increased, sta6 may be able to utilize the substantially elevated pool of malonyl CoA. Transcriptomic studies of C. reinhardtii have shown a significant increase in the content of acyl transferases such as acyl CoA/diacylglycerol acyltransferase (DGAT) and phospholipid/diacylglycerol acyltransferase, and suggested the involvement of transcription factor NRR1 in potentially regulating the expression of certain diacylglycerol acyltransferase genes and triacylglycerol accumulation (Boyle et al., 2012). The activities of these enzymes in particular may have to be increased to achieve improved lipid yields from malonyl CoA in sta6.

In conclusion, our data indicate that, in C. reinhardtii, starch biosynthesis plays a critical role in regulating multiple functions, largely through accumulation/utilization of redox and adenylate cofactors. Future research efforts are required to examine mechanisms to effectively leverage the greatly increased malonyl CoA levels for enhanced lipid biosynthesis, which, if successful, may allow higher NADPH re-oxidation rates and restore photosynthetic productivity.

EXPERIMENTAL PROCEDURES

Strains and culture conditions

Chlamydomonas strains CC-4349 cw15 mt+ (Goodenough 330A, referred to as cw15) and cw15 sta6 (BAFJ5, referred to as sta6) (Zabawinski et al., 2001; Wang et al., 2009) were obtained from...
the Chlamydomonas Resource Center (University of Minnesota, St Paul, MN). cw15 was chosen as the control strain as it was thought to be the clonal parent to sta6. However, during writing of this manuscript, Blaby et al. (2013) showed that it is not the original parent, as they do not share the same mating type. However, cw15 was retained as the control strain in this study because (i) it has an intact starch biosynthetic pathway (Blaby et al., 2013), (ii) like sta6, it lacks a cell wall, and (iii) it does not require arginine for growth, making it suitable for autotrophic photosynthetic studies. Liquid cultures were grown in phosphate-buffered Sueoka’s high-salt (HS) medium supplemented with 5 mM NaHCO3 and 9.4 mM NH4Cl (Harris and Stern, 2009). The cultures were maintained on an orbital shaker (100 rpm) under continuous illumination (100 μE m−2 sec−1, photosynthetically active radiation) at 25°C. For growth rate measurements, flasks were inoculated at a density of approximately 6 × 105 cells ml−1, and cell density was measured daily using a hemocytometer. Growth data were fitted to a Gompertz function (Zwietering and cell density was measured daily using a hemocytometer. Growth data were fitted to a Gompertz function (Zwietering et al., 1990) to calculate the specific growth rate and doubling time. For all other characterizations, cells were grown in semi-continuous cultures. Pre-cultures were grown to approximately 15 × 106 cells ml−1 and then resuspended at 2.5 × 106 cells ml−1 in fresh HS medium. Samples were taken immediately after resuspension (0 h) and at 96 h.

Chlorophyll measurements
Chlorophyll concentration was determined spectrophotometrically by methanol extraction using extinction coefficients from Porra et al. (1989).

In vivo measurements of oxygen evolution
Light-saturated OERs were measured using a membrane-covered Clark O2 electrode (Hansatech, http://hansatech-instruments.com/) at 25°C in HS medium containing 5 mM NaHCO3. Samples were illuminated with red light. Light-dependent respiration was measured within 30 sec after onset of darkness, and subtracted from the light-dependent OER to obtain the gross OER. Appropriate electron acceptors and inhibitors were added directly to the sample chamber. Titration curves for the PSII and PSI electron acceptors and inhibitors were added directly to the sample chamber. Time-resolved OERs were measured using a home-built bare platinum electrode controlled by Labview software (Zakrzhevskii et al., 1978), which allows kinetic resolution of the transit time to electron acceptor pools downstream of PSII (NADP+ and CO2). Cells were concentrated to 20 μg ml−1 chlorophyll, and 8 μl aliquots were loaded onto the 8 mm diameter electrode, dark-incubated for 15 min, and illuminated with red light (400 μE m−2 sec−1). The full kinetics of O2 evolution were recorded over 2 min. The O2 signals thus generated reflect the size of the electron acceptor pools and the transit times required to fill them (Zakrzhevskii et al., 1978). The O2 peak associated with reduction of the PQ pool was isolated by subtracting the slower-filling pools by linear extrapolation to the baseline, and the total O2 evolved was determined from the current using Faraday’s law of electrolysis.

Pyridine nucleotide fluorescence kinetics
NAD(P)H fluorescence induction kinetics were measured in whole cells using a home-built instrument similar to one described previously (Mi et al., 2000). A UV-LED (365 nm wavelength) in combination with an optical filter (2 mm UG11, Schott, http://www.schott.com) was used as the excitation light source for NAD(P)H fluorescence. An NMR tube with an outer diameter of 5 mm was used as the sample holder, and the detection and excitation pathways were perpendicular. A multifunctional PCI-6036E DAC board (National Instruments, http://www.ni.com/) was used for data acquisition.

Metabolite extraction and analysis
Cultures (8.8-11 × 106 cells ml−1) were quenched in a 60:40 MeOH/water mixture, and centrifuged at 3500 g for 2 min at 0°C. The supernatant was discarded and the pellet was resuspended in a ice-cold 60:40 methanol/water mixture. The suspension was sonicated for 1 min, and then incubated at −20°C for 20 min. Then the samples were centrifuged for 5 min at 5000 g at 0°C. The supernatant was vacuum-dried using a Centrif Vap benchtop vacuum concentrator (Labconoco, www.labconco.com), and the resulting pellet was resuspended in LC-MS grade water. Samples were injected into an Agilent 6490 QQQ mass analyzer coupled to an HPLC (Agilent Technologies, www.agilent.com) for metabolite analysis (Kenchappa et al., 2013). Concentrations of each metabolite per cell were calculated using the calibration curve. Fold changes were calculated as a ratio of the concentration of metabolites under LL and HL conditions.
Lipid, glucose, and protein analysis

Total lipids were converted to fatty acids for GC-FID analysis as described previously (Radakovits et al., 2012), and analyzed using an Agilent 7890A gas chromatograph and DB5 ms column with flame ionization detection. Total sugar was analyzed using the anthrone assay (Morris, 1948; Meuser et al., 2012). Protein levels were determined using a DC protein assay kit (Bio-Rad, www.bio-rad.com) according to the manufacturer’s instructions.

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SUPPORTING INFORMATION

Additional Supporting Information may be found in the online version of this article.

Figure S1. Percentage difference between OER and ETR

Figure S2. Dark/light/dark induction transients of blue-green fluorescence in *Chlamydomonas reinhardtii*

Figure S3. Titration curves for 2,4-dichloro-p-benzoquinone and N-dimethyl-4-nitrosoaniline.

REFERENCES


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Photosynthetic metabolism in starchless alga 959


